

ERYTHRINA STUDIES. PART 1. NOVEL ANTIBACTERIAL
FLAVANONES FROM ERYTHRINA SIGMOIDEA

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Summary : Two new flavanones, sigmoidin A and B have been isolated from the Cameroonian medicinal plant Erythrina sigmoidea. The two compounds exhibit significant antibacterial activity against Gram-positive bacteria.

Erythrina alkaloids have attracted attention for decades¹. The neutral components of this genus have, on the other hand, received very little attention. However, recently, Nakanishi and co-workers² isolated from E. abyssinica several flavonoids which displayed noteworthy biological activities. In continuation of our studies on Cameroonian medicinal plants³ we have investigated the widely used folk medicinal plant, Erythrina sigmoidea⁴, and now report the characterisation of two novel antibacterial flavanones from the chloroform extract.

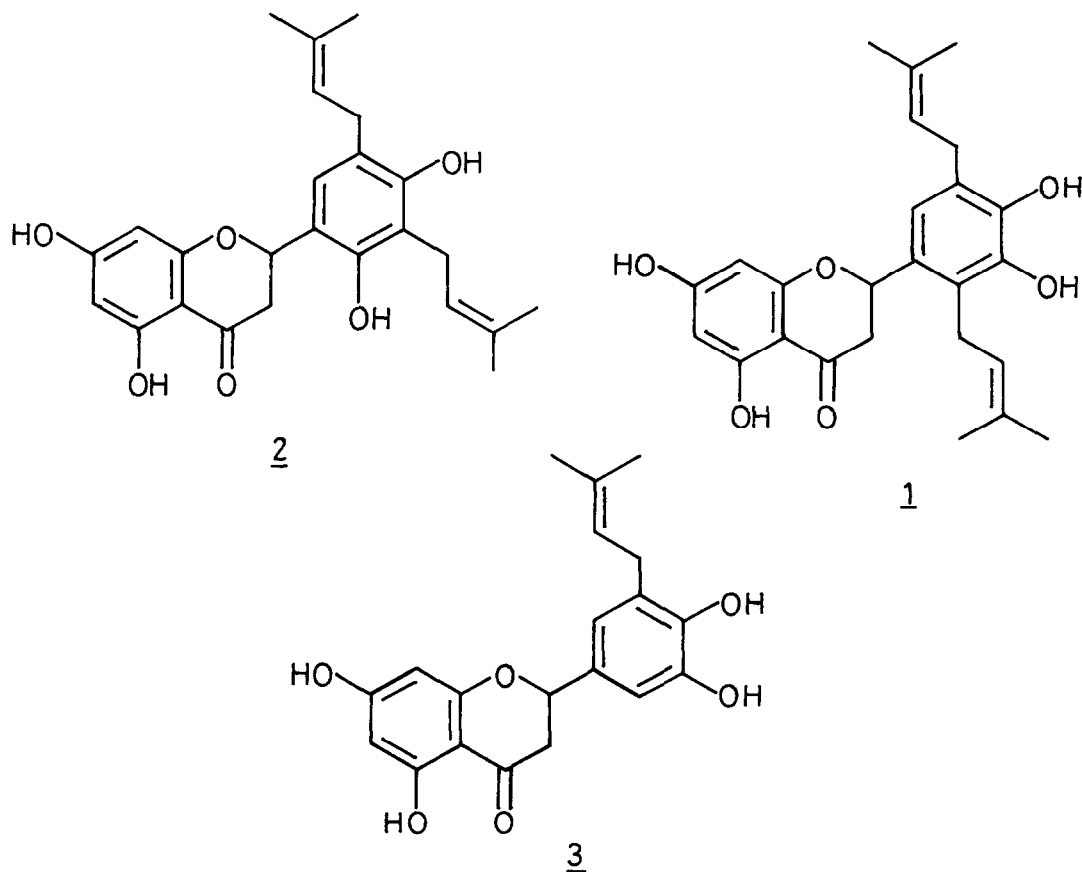
The first compound for which we propose the name sigmoidin A, 1 m.p. 180-181°, was shown to have the composition C₂₅H₂₈O₆ by elemental analysis and mass spectral data (M⁺ at m/z 424). Colour tests with FeCl₃ (green) and magnesium-concentrated hydrochloric acid (pink) together with the u.v. spectral data { $\lambda_{\max}^{\text{MeOH}}$: 288nm (ϵ 12 000), $\lambda_{\max}^{\text{MeOH} + \text{NaOMe}}$ 323nm (20 900) ; $\lambda_{\max}^{\text{MeOH} + \text{NaOAc}}$: 325nm (18 800) and $\lambda_{\max}^{\text{MeOH} + \text{AlCl}_3}$: 309nm (15 400) (unchanged on addition of HCl)} indicated that sigmoidin A was a flavanone bearing at least two hydroxy-groups⁵. Further indication of this skeleton came from the i.r. spectrum of 1 which exhibited strong absorptions at ν_{\max} (KBr) 3500 (free OH), 3300-3100 (bonded OH) and 1640 cm⁻¹ (chelated C=O). The trimethyl ether of 1, (M⁺ 466) from diazomethane methylation gave a green colour with FeCl₃ confirming the

presence of a chelated hydroxy-group while acetylation with acetic anhydride-pyridine yielded a tetraacetate which did not respond to the FeCl_3 test. Thus sigmoidin A contains four hydroxy-groups. The bathochromic shifts in the u.v. spectrum of 1 induced by NaOMe, NaOAc and AlCl_3 are consistent with its formulation as a flavanone hydroxylated at C-5 and C-7⁵. The ^1H . n.m.r. spectrum (DMSO- d_6) of 1 confirmed its 5,7-dihydroxylated nature and showed resonances for the characteristic flavanone 2H-3 and H-2 protons at δ 2.80 (1H, dd, $J=17\text{Hz}$ and 4Hz , H-3), 3.16 (1H, dd, $J=17\text{Hz}$ and 4Hz , H-3) and 5.33 (1H, m, H-2), and four D_2O -exchangeable signals for four hydroxy-groups at δ 7.85, 8.05, 10.40 and 12.01. Further signals at δ 1.67 (12H, bs), 3.24 (4H, d, $J=7.5\text{Hz}$) and 5.28 (2H, ill-defined t) indicated the presence of two 3-methylbut-2-enyl (prenyl) groups. Resonances for three aromatic protons were also observed at δ 5.84 (2H, s) and 6.70 (1H, s). The former were assigned to H-6 and H-8 in accordance with chemical shift data recorded for these two protons² while the latter obviously arose from ring-B. It thus follows that sigmoidin A is a 5,7-dihydroxyflavanone bearing four substituents : two hydroxy and two prenyl groups in ring B whose relative positions had to be determined. On biogenetic grounds⁶, it was assumed that there would be hydroxylation at C-4'. Furthermore, in sigmoidin tetracetate the H-6 and H-8 singlet at δ 5.84 was split into two symmetric doublets ($J=2.5\text{Hz}$) and shifted to δ 6.65 and 6.8 while the ring-B proton singlet remained almost unchanged at δ 6.68. The absence of any downfield shift for this proton on acetylation indicated that it was not ortho to a hydroxy-group^{7,8}. Hence two possible structures 1 and 2 could be assigned to sigmoidin A. Mild formic acid-catalysed cyclisation of sigmoidin A furnished a single dichromano-derivative which lacked prenyl absorption in its ^1H n.m.r. spectrum. This result ruled out the possibility of 2 which must give more than two chromano-derivatives in this reaction. Sigmoidin A could therefore be formulated as 1.

The major compound, sigmoidin B, 3, $\text{C}_{20}\text{H}_{20}\text{O}_6$, (M^+ 356) had m.p. 217-218°. Preliminary colour tests showed that 3 was also a hydroxylated flavanone. The i.r. spectrum of 3 (ν_{max} (KBr) 3600, 3450-3150 and 1635 cm^{-1} (chelated C=O) and the u.v. spectrum ($\lambda_{\text{max}}^{\text{MeOH}}$: 288 nm (ϵ 12900) ; $\lambda_{\text{max}}^{\text{MeOH} + \text{NaOMe}}$: 325 nm (21800) ; $\lambda_{\text{max}}^{\text{MeOH} + \text{NaOAc}}$: 323 (19600) and $\lambda_{\text{max}}^{\text{MeOH} + \text{AlCl}_3}$: 309 nm (16000) unchanged on addition of HCl) further indicated that 3 had the same 5,7-dihydroxylation pattern as sigmoidin A⁵. The ^1H n.m.r. spectrum of sigmoidin B (3) was well-resolved and showed besides the characteristic 2H-3 and H-2 resonances (δ 2.84 (1H, dd, $J=17$ and 4Hz , H-3), 3.20 (1H, dd, $J=17$ and 4Hz , H-3) and δ 5.40 (1H, ill-defined t, H-2)), signals for one chelated hydroxy-group at δ 12.02 (1H, s, exchangeable with D_2O) and one 3-methylbut-2-enyl group. Resonances for four aromatic protons were also observed at δ 5.86 (2H, s), 6.61 (1H, d, $J=2\text{Hz}$) and 6.75 (1H, d, $J=2\text{Hz}$). The former were assigned to H-6 and H-8² while the other

two obviously arose from ring-B. In agreement with the above spectral data 3 on diazomethane methylation furnished a trimethyl ether and on acetylation, a tetra-acetate. Sigmoidin B 3 therefore differed from sigmoidin A (1) only by having one prenyl group less in ring-B. The establishment of the structure of sigmoidin B thus resolved itself into one of determining the positions of the prenyl and the two hydroxy groups. One of the hydroxy-groups was assigned to C-4' on biogenetic considerations⁶. Treatment of sigmoidin B with 98% formic acid gave a single chromano-derivative (M^+ 356) indicating that the prenyl group was flanked by a single hydroxy-group. The existence of two meta-coupled ring-B protons (¹H n.m.r spectrum vide supra) coupled with the above evidence uniquely defined the B-ring substitution pattern as 3',4'-dihydroxy-5'-(3"-methylbut-2"-enyl). Hence sigmoidin B has the novel structure 5,7,3'4'-tetrahydroxy-5'-(3"-methylbut-2"-enyl)-flavanone 3.

Preliminary antibacterial tests show that sigmoidin A and B strongly inhibited Staphylococcus aureus and Bacillus subtilis at 50 p.p.m. The exact MIC will be published in detail elsewhere.



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REFERENCES

1. S.F. Dyke and D. Quessy in "The Alkaloids : Chemistry and Physiology". Vol. 18 (Edited by R.G.A. Rodrigo), p. 1, Academic Press Inc., New York (1981) and references cited therein.
2. V.S. Kamat, F.Y. Chuo, I. Kubo and K. Nakanishi, Heterocycles, 15, 1163 (1981).
3. J.F. Ayafor, B.L. Sondengam, M.N. Bilon, E. Tsamo, S.F. Kimbu and J.I. Okogun, J. Natural Products, 45, 714 (1982) and references cited therein.
4. A. Bouquet and M. Debray, "Plantes Medicinales de la Côte d'Ivoire", ORSTOM, Paris, 136 (1974).
5. T.J. Mabry, K.R. Markham and M.B. Thomas, "The systematic Identification of Flavonoids", p. 165, Springer Verlag, New York (1970).
6. B.A. Bohm in "The Flavonoids" (Edited by J.B. Harbone, T.J. Mabry and H. Mabry), p. 560, Chapman and Hall LTD, London (1975).
7. N.W. Preston, Phytochemistry, 16, 143 (1977).
8. S. Bhanumati, S.C. Chhabra and S.R. Gupta, Phytochemistry, 18, 1254 (1979).

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